

## 5-(Anthracen-9-yl)-3-(4-nitrophenyl)-1-phenyl-4,5-dihydro-1H-pyrazole

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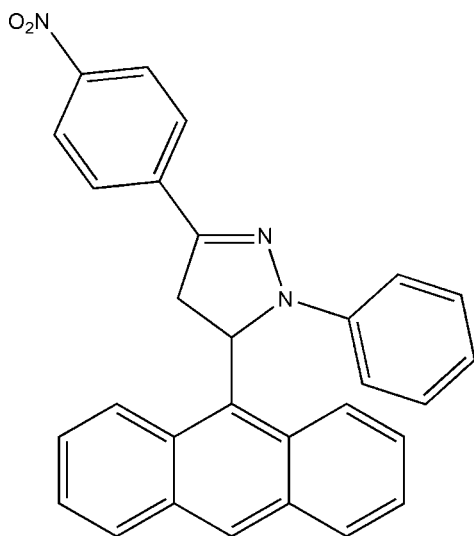
Received 25 September 2010; accepted 29 September 2010

 Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.095; data-to-parameter ratio = 6.6.

In the title compound,  $\text{C}_{29}\text{H}_{21}\text{N}_3\text{O}_2$ , the five-membered pyrazoline ring is nearly planar, the maximum deviation being 0.037 (3) Å. The anthracene ring system is approximately perpendicular to the central pyrazoline ring, making a dihedral angle of 86.55 (16)°, whereas the two attached benzene rings are oriented at smaller dihedral angles of 12.9 (2) and 14.7 (2)° with respect to the pyrazoline ring. An intramolecular  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bond is observed.

### Related literature

For applications of pyrazoline derivatives, see: Shaharyar *et al.* (2006); Christoph *et al.* (2003); Parmar *et al.* (1974); Prasad *et al.* (2005). For a related pyrazoline compound, see: Krishna *et al.* (1999).



### Experimental

#### Crystal data

$\text{C}_{29}\text{H}_{21}\text{N}_3\text{O}_2$	$V = 2159.6$ (7) Å <sup>3</sup>
$M_r = 443.49$	$Z = 4$
Orthorhombic, $Pca2_1$	Mo $K\alpha$ radiation
$a = 23.023$ (5) Å	$\mu = 0.09$ mm <sup>-1</sup>
$b = 10.195$ (2) Å	$T = 293$ K
$c = 9.2005$ (18) Å	$0.20 \times 0.20 \times 0.20$ mm

#### Data collection

Rigaku Mercury2 diffractometer	1675 reflections with $I > 2\sigma(I)$
17812 measured reflections	$R_{\text{int}} = 0.068$
2023 independent reflections	

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$	2 restraints
$wR(F^2) = 0.095$	H-atom parameters constrained
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.14$ e Å <sup>-3</sup>
2023 reflections	$\Delta\rho_{\text{min}} = -0.16$ e Å <sup>-3</sup>
308 parameters	

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}1-\text{H}1A\cdots\text{N}2$	0.93	2.59	3.422 (5)	150

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5037).

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**supplementary materials**

*Acta Cryst.* (2010). E66, o2786 [ doi:10.1107/S1600536810038912 ]

## 5-(Anthracen-9-yl)-3-(4-nitrophenyl)-1-phenyl-4,5-dihydro-1H-pyrazole

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### Comment

The derivatives of pyrazoline are mostly used in medicine, such as antibacterial (Shaharyar *et al.*, 2006), antidepressant (Prasad *et al.*, 2005) and anticonvulsant (Parmar *et al.*, 1974), furthermore they also have application in cell biological study due to their simple structure and favorable photophysical properties (Christoph *et al.*, 2003). Here we report the structure of the title compound (I), a new derivative of pyrazoline.

In the pyrazoline ring, the bond length of C17=N2 [1.287 (4) Å] indicates a to the normal C=N bond (1.28 Å), while the N2—N3 distance [1.366 (4) Å] agrees with the expected values (Krishna *et al.*, 1999). The mean plane of anthryl ring makes dihedral angles of 84.98 (9) and 82.81 (8)°, with the benzene ring and 4-nitrophenyl group, respectively.

### Experimental

3-(9-Anthryl)-1-(4-nitrophenylprop)-2-en-1-one (3 mmol, 1.0 g) and phenylhydrazine (6.5 mmol, 0.7 g) were dissolved in 10 ml acetic acid. The mixture was stirred for 8 h at refluxing temperature to give red solid. The product was isolated and recrystallized from ethanol/ethyl acetate (1:1 v/v) mixed solution, red single-crystal of (1) was obtained.

### Refinement

H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . As no significant anomalous scatterings, Friedel pairs were merged.

### Figures

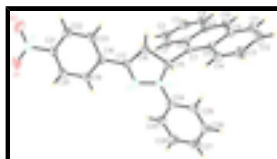


Fig. 1. Perspective structure of the title compound, showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

## 5-(Anthracen-9-yl)-3-(4-nitrophenyl)-1-phenyl-4,5-dihydro-1H-pyrazole

### Crystal data

C<sub>29</sub>H<sub>21</sub>N<sub>3</sub>O<sub>2</sub>

$M_r = 443.49$

Orthorhombic, *Pca*2<sub>1</sub>

Hall symbol: P 2c -2ac

$a = 23.023 (5) \text{ \AA}$

$F(000) = 928$

$D_x = 1.364 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 2023 reflections

$\theta = 2.6\text{--}25.0^\circ$

# supplementary materials

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$b = 10.195 (2) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 9.2005 (18) \text{ \AA}$	$T = 293 \text{ K}$
$V = 2159.6 (7) \text{ \AA}^3$	Prism, red
$Z = 4$	$0.20 \times 0.20 \times 0.20 \text{ mm}$

## Data collection

Rigaku Mercury2 diffractometer	1675 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube graphite	$R_{\text{int}} = 0.068$
Detector resolution: $13.6612 \text{ pixels mm}^{-1}$	$\theta_{\text{max}} = 25.0^\circ$ , $\theta_{\text{min}} = 3.1^\circ$
CCD_Profile_fitting scans	$h = -27 \rightarrow 27$
17812 measured reflections	$k = -12 \rightarrow 12$
2023 independent reflections	$l = -10 \rightarrow 10$

## Refinement

Refinement on $F^2$	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.044$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.095$	H-atom parameters constrained
$S = 1.08$	$w = 1/[\sigma^2(F_o^2) + (0.0418P)^2 + 0.3731P]$
2023 reflections	where $P = (F_o^2 + 2F_c^2)/3$
308 parameters	$(\Delta/\sigma)_{\text{max}} < 0.001$
2 restraints	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
N2	0.34781 (11)	0.8437 (3)	0.1581 (3)	0.0417 (7)
C18	0.33011 (15)	0.6798 (3)	-0.0211 (4)	0.0400 (8)
C6	0.48920 (14)	0.7455 (3)	0.3722 (4)	0.0392 (8)

C7	0.49683 (14)	0.8475 (3)	0.2692 (4)	0.0380 (8)
C5	0.53459 (15)	0.7161 (3)	0.4735 (4)	0.0451 (9)
C17	0.36872 (14)	0.7631 (4)	0.0641 (4)	0.0410 (9)
C21	0.25650 (16)	0.5236 (3)	-0.1831 (4)	0.0437 (9)
C9	0.59566 (15)	0.8817 (4)	0.3653 (4)	0.0461 (9)
O1	0.16551 (13)	0.4657 (3)	-0.2656 (4)	0.0673 (9)
C8	0.55013 (14)	0.9175 (3)	0.2655 (4)	0.0394 (9)
C15	0.44934 (13)	0.8788 (3)	0.1582 (4)	0.0408 (8)
H15A	0.4619	0.9549	0.1013	0.049*
N3	0.39237 (12)	0.9097 (3)	0.2234 (4)	0.0434 (7)
C29	0.42219 (16)	1.1131 (3)	0.3361 (4)	0.0437 (9)
H29A	0.4595	1.1022	0.2991	0.052*
C1	0.43743 (16)	0.6694 (3)	0.3833 (4)	0.0464 (9)
H1A	0.4074	0.6842	0.3178	0.056*
C27	0.35497 (16)	1.2377 (4)	0.4779 (5)	0.0594 (11)
H27A	0.3468	1.3096	0.5367	0.071*
C11	0.65023 (17)	0.9509 (4)	0.3584 (5)	0.0613 (12)
H11A	0.6801	0.9270	0.4211	0.074*
C14	0.56258 (15)	1.0242 (3)	0.1698 (5)	0.0482 (9)
H14A	0.5339	1.0515	0.1053	0.058*
C24	0.37940 (15)	1.0216 (3)	0.3050 (4)	0.0409 (9)
C10	0.58675 (16)	0.7836 (4)	0.4662 (4)	0.0506 (10)
H10A	0.6164	0.7624	0.5307	0.061*
N1	0.21739 (15)	0.4423 (3)	-0.2714 (4)	0.0558 (9)
C23	0.35171 (16)	0.5779 (4)	-0.1053 (5)	0.0546 (11)
H23A	0.3915	0.5628	-0.1076	0.066*
C2	0.43091 (17)	0.5761 (4)	0.4868 (5)	0.0564 (11)
H2A	0.3967	0.5277	0.4904	0.068*
C20	0.23398 (15)	0.6237 (4)	-0.1030 (4)	0.0476 (10)
H20A	0.1942	0.6399	-0.1041	0.057*
O2	0.23825 (14)	0.3550 (3)	-0.3453 (4)	0.0887 (11)
C28	0.40961 (17)	1.2200 (4)	0.4218 (5)	0.0509 (10)
H28A	0.4385	1.2810	0.4419	0.061*
C12	0.65894 (18)	1.0493 (4)	0.2635 (6)	0.0689 (13)
H12A	0.6946	1.0921	0.2605	0.083*
C16	0.43391 (13)	0.7676 (4)	0.0517 (4)	0.0453 (9)
H16A	0.4461	0.7885	-0.0466	0.054*
H16B	0.4514	0.6852	0.0810	0.054*
C13	0.61422 (17)	1.0873 (4)	0.1690 (5)	0.0577 (11)
H13A	0.6201	1.1567	0.1051	0.069*
C3	0.47497 (19)	0.5511 (4)	0.5891 (5)	0.0590 (11)
H3A	0.4693	0.4889	0.6617	0.071*
C25	0.32422 (17)	1.0409 (4)	0.3603 (5)	0.0566 (11)
H25A	0.2949	0.9812	0.3393	0.068*
C19	0.27035 (15)	0.7011 (3)	-0.0203 (4)	0.0474 (9)
H19A	0.2549	0.7680	0.0365	0.057*
C4	0.52555 (19)	0.6179 (4)	0.5811 (5)	0.0584 (11)
H4A	0.5550	0.5993	0.6471	0.070*
C22	0.31528 (17)	0.4985 (4)	-0.1856 (5)	0.0572 (11)

## supplementary materials

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H22A	0.3301	0.4294	-0.2403	0.069*
C26	0.31263 (17)	1.1481 (4)	0.4460 (6)	0.0679 (13)
H26A	0.2754	1.1600	0.4829	0.082*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N2	0.0337 (16)	0.0427 (16)	0.0487 (18)	0.0034 (14)	-0.0089 (15)	-0.0072 (16)
C18	0.0364 (18)	0.0455 (19)	0.038 (2)	-0.0010 (16)	-0.0049 (17)	-0.0013 (18)
C6	0.0333 (19)	0.0358 (18)	0.048 (2)	0.0078 (15)	-0.0037 (16)	-0.0067 (18)
C7	0.0317 (18)	0.0382 (18)	0.044 (2)	0.0055 (15)	-0.0054 (16)	-0.0067 (18)
C5	0.046 (2)	0.044 (2)	0.045 (2)	0.0123 (17)	-0.0091 (19)	-0.005 (2)
C17	0.0322 (18)	0.049 (2)	0.041 (2)	0.0033 (16)	-0.0077 (16)	-0.0047 (19)
C21	0.040 (2)	0.052 (2)	0.0393 (19)	-0.0063 (18)	-0.0050 (18)	-0.0046 (19)
C9	0.032 (2)	0.054 (2)	0.052 (2)	0.0026 (17)	-0.0061 (17)	-0.014 (2)
O1	0.0474 (17)	0.081 (2)	0.074 (2)	-0.0142 (15)	-0.0129 (17)	-0.0069 (18)
C8	0.0347 (19)	0.042 (2)	0.042 (2)	0.0047 (15)	-0.0029 (16)	-0.0111 (18)
C15	0.0331 (18)	0.044 (2)	0.045 (2)	0.0027 (15)	-0.0080 (17)	-0.0037 (19)
N3	0.0310 (15)	0.0459 (17)	0.0534 (18)	0.0040 (14)	-0.0084 (15)	-0.0110 (16)
C29	0.041 (2)	0.041 (2)	0.049 (2)	0.0030 (17)	-0.0015 (18)	-0.0050 (19)
C1	0.041 (2)	0.042 (2)	0.056 (2)	0.0045 (17)	-0.0084 (19)	-0.001 (2)
C27	0.055 (3)	0.052 (2)	0.071 (3)	0.002 (2)	0.005 (2)	-0.022 (2)
C11	0.038 (2)	0.074 (3)	0.072 (3)	0.002 (2)	-0.012 (2)	-0.015 (3)
C14	0.040 (2)	0.050 (2)	0.055 (2)	0.0007 (18)	0.0018 (19)	-0.004 (2)
C24	0.040 (2)	0.0387 (19)	0.044 (2)	0.0035 (16)	-0.0075 (17)	-0.0030 (18)
C10	0.038 (2)	0.060 (2)	0.054 (3)	0.0114 (19)	-0.014 (2)	-0.012 (2)
N1	0.054 (2)	0.064 (2)	0.050 (2)	-0.0117 (18)	-0.0066 (18)	-0.003 (2)
C23	0.035 (2)	0.076 (3)	0.053 (2)	0.0052 (19)	-0.0053 (19)	-0.021 (2)
C2	0.054 (2)	0.045 (2)	0.071 (3)	0.0012 (19)	0.001 (2)	0.003 (2)
C20	0.0313 (19)	0.055 (2)	0.056 (2)	0.0010 (17)	-0.0054 (19)	-0.002 (2)
O2	0.075 (2)	0.097 (2)	0.094 (3)	-0.007 (2)	-0.001 (2)	-0.055 (2)
C28	0.048 (2)	0.041 (2)	0.064 (3)	0.0001 (18)	-0.004 (2)	-0.005 (2)
C12	0.048 (3)	0.071 (3)	0.088 (4)	-0.017 (2)	0.002 (3)	-0.012 (3)
C16	0.0333 (19)	0.059 (2)	0.044 (2)	0.0019 (17)	-0.0053 (17)	-0.010 (2)
C13	0.046 (2)	0.060 (3)	0.067 (3)	-0.009 (2)	0.005 (2)	-0.007 (2)
C3	0.071 (3)	0.047 (2)	0.059 (3)	0.013 (2)	0.001 (2)	0.011 (2)
C25	0.039 (2)	0.056 (2)	0.074 (3)	-0.0022 (18)	0.004 (2)	-0.019 (2)
C19	0.040 (2)	0.047 (2)	0.055 (2)	0.0028 (17)	-0.0064 (19)	-0.012 (2)
C4	0.061 (3)	0.056 (2)	0.058 (3)	0.018 (2)	-0.011 (2)	0.005 (2)
C22	0.048 (2)	0.070 (3)	0.053 (2)	0.006 (2)	-0.002 (2)	-0.023 (2)
C26	0.044 (2)	0.073 (3)	0.087 (4)	0.005 (2)	0.013 (2)	-0.029 (3)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

N2—C17	1.287 (4)	C27—C26	1.368 (5)
N2—N3	1.366 (4)	C27—C28	1.372 (5)
C18—C23	1.388 (5)	C27—H27A	0.9300
C18—C19	1.393 (5)	C11—C12	1.345 (6)
C18—C17	1.458 (5)	C11—H11A	0.9300

C6—C7	1.417 (5)	C14—C13	1.352 (5)
C6—C1	1.426 (5)	C14—H14A	0.9300
C6—C5	1.432 (5)	C24—C25	1.382 (5)
C7—C8	1.420 (5)	C10—H10A	0.9300
C7—C15	1.530 (5)	N1—O2	1.219 (4)
C5—C10	1.386 (5)	C23—C22	1.380 (5)
C5—C4	1.423 (6)	C23—H23A	0.9300
C17—C16	1.506 (5)	C2—C3	1.407 (6)
C21—C20	1.361 (5)	C2—H2A	0.9300
C21—C22	1.378 (5)	C20—C19	1.379 (5)
C21—N1	1.469 (5)	C20—H20A	0.9300
C9—C10	1.381 (6)	C28—H28A	0.9300
C9—C8	1.440 (5)	C12—C13	1.402 (6)
C9—C11	1.442 (5)	C12—H12A	0.9300
O1—N1	1.219 (4)	C16—H16A	0.9700
C8—C14	1.429 (5)	C16—H16B	0.9700
C15—N3	1.476 (4)	C13—H13A	0.9300
C15—C16	1.540 (5)	C3—C4	1.351 (6)
C15—H15A	0.9800	C3—H3A	0.9300
N3—C24	1.398 (4)	C25—C26	1.374 (5)
C29—C28	1.375 (5)	C25—H25A	0.9300
C29—C24	1.387 (5)	C19—H19A	0.9300
C29—H29A	0.9300	C4—H4A	0.9300
C1—C2	1.354 (5)	C22—H22A	0.9300
C1—H1A	0.9300	C26—H26A	0.9300
C17—N2—N3	109.2 (3)	C25—C24—N3	120.7 (3)
C23—C18—C19	118.3 (3)	C29—C24—N3	120.5 (3)
C23—C18—C17	121.2 (3)	C9—C10—C5	121.4 (3)
C19—C18—C17	120.6 (3)	C9—C10—H10A	119.3
C7—C6—C1	123.4 (3)	C5—C10—H10A	119.3
C7—C6—C5	119.9 (3)	O2—N1—O1	123.6 (3)
C1—C6—C5	116.7 (3)	O2—N1—C21	118.6 (3)
C6—C7—C8	119.5 (3)	O1—N1—C21	117.8 (3)
C6—C7—C15	120.7 (3)	C22—C23—C18	121.3 (3)
C8—C7—C15	119.8 (3)	C22—C23—H23A	119.3
C10—C5—C4	120.7 (4)	C18—C23—H23A	119.3
C10—C5—C6	119.8 (4)	C1—C2—C3	121.2 (4)
C4—C5—C6	119.6 (3)	C1—C2—H2A	119.4
N2—C17—C18	120.4 (3)	C3—C2—H2A	119.4
N2—C17—C16	113.8 (3)	C21—C20—C19	119.7 (3)
C18—C17—C16	125.8 (3)	C21—C20—H20A	120.1
C20—C21—C22	121.5 (3)	C19—C20—H20A	120.1
C20—C21—N1	119.3 (3)	C27—C28—C29	120.9 (4)
C22—C21—N1	119.2 (3)	C27—C28—H28A	119.6
C10—C9—C8	120.3 (3)	C29—C28—H28A	119.6
C10—C9—C11	120.9 (4)	C11—C12—C13	120.0 (4)
C8—C9—C11	118.8 (4)	C11—C12—H12A	120.0
C7—C8—C14	124.8 (3)	C13—C12—H12A	120.0
C7—C8—C9	119.1 (3)	C17—C16—C15	101.8 (3)

## supplementary materials

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C14—C8—C9	116.1 (3)	C17—C16—H16A	111.4
N3—C15—C7	114.1 (3)	C15—C16—H16A	111.4
N3—C15—C16	102.2 (3)	C17—C16—H16B	111.4
C7—C15—C16	115.9 (3)	C15—C16—H16B	111.4
N3—C15—H15A	108.1	H16A—C16—H16B	109.3
C7—C15—H15A	108.1	C14—C13—C12	120.7 (4)
C16—C15—H15A	108.1	C14—C13—H13A	119.6
N2—N3—C24	118.5 (3)	C12—C13—H13A	119.6
N2—N3—C15	112.5 (3)	C4—C3—C2	119.6 (4)
C24—N3—C15	125.6 (3)	C4—C3—H3A	120.2
C28—C29—C24	120.1 (3)	C2—C3—H3A	120.2
C28—C29—H29A	120.0	C26—C25—C24	120.2 (4)
C24—C29—H29A	120.0	C26—C25—H25A	119.9
C2—C1—C6	121.7 (3)	C24—C25—H25A	119.9
C2—C1—H1A	119.1	C20—C19—C18	120.5 (3)
C6—C1—H1A	119.1	C20—C19—H19A	119.7
C26—C27—C28	119.0 (4)	C18—C19—H19A	119.7
C26—C27—H27A	120.5	C3—C4—C5	121.2 (4)
C28—C27—H27A	120.5	C3—C4—H4A	119.4
C12—C11—C9	121.5 (4)	C5—C4—H4A	119.4
C12—C11—H11A	119.2	C21—C22—C23	118.6 (4)
C9—C11—H11A	119.2	C21—C22—H22A	120.7
C13—C14—C8	122.8 (4)	C23—C22—H22A	120.7
C13—C14—H14A	118.6	C27—C26—C25	121.1 (4)
C8—C14—H14A	118.6	C27—C26—H26A	119.5
C25—C24—C29	118.8 (3)	C25—C26—H26A	119.5

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1A $\cdots$ N2	0.93	2.59	3.422 (5)	150



Fig. 1

